

CONFORMATIONALLY SELECTIVE TRANSANNULAR CYCLIZATIONS OF HUMULENE 9,10-EPOXIDE.
SYNTHESIS OF THE TWO SKELETALLY DIFFERENT CYCLOHUMULANOIDS:
DL-BICYCLOHUMULENONE AND DL-AFRICANOL

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Summary — Two cyclohumulanoids, dl-bicyclohumulenone (**4**) and dl-africanol (**7**) were synthesized through newly developed conformationally selective transannular cyclization of humulene 9,10-epoxide (**2**). The epoxide **2** was converted to a bicyclohumulenediol diacetate **3a** in 70% yield by treatment with $\text{BF}_3 \cdot \text{OEt}_2 \cdot \text{Ac}_2\text{O}$, while treatment of **2** with trimethylsilyl trifluoromethanesulfonate gave an africen-10-ol (**5a** and **5b**) in 80% yield. The two intermediates **3a** and **5a** furnished the natural products **4** and **7** in 30 and 8% yield from **2** respectively.

Previously we reported¹⁾ that humulene (**1**) is in equilibrium with its two stable CT and CC conformers and suggested that illudoids²⁾ and hirsutanoids might be biosynthetically derived from the CT and CC conformers respectively.¹⁾ If one of the two stable conformers at equilibrium can be selectively converted to the corresponding biosynthetically derived cyclization products,³⁾ the process will make our simulation of cyclohumulanoid biosynthesis⁴⁾ complete. Moreover, cyclization of a designated conformer *via* a conformationally selective transannular cyclization constitutes a new strategy in synthetic organic chemistry.

Most cyclohumulanoids are considered to stem from 9,10-dihydrohumulen-9-yl cation.⁵⁾ However, selective protonation of the 9,10-double bond of humulene is so difficult⁶⁾ that we have attempted generation of the cation by cleavage of the epoxide ring of humulene 9,10-epoxide (**2**).⁷⁾ Since stable conformations of humulene epoxides are known to be very similar to those of the original olefin,³⁾ epoxide **2** undoubtedly exists as a mixture of CT and CC conformations at equilibrium.

The epoxide **2** upon treatment with $\text{BF}_3 \cdot \text{OEt}_2$ (0.8 eq) in Ac_2O at -50° followed by 1 hr at -20°C gave rise to bicyclohumulenediol diacetate **3a**⁹⁾ (mp $140-2^\circ$) in 70% yield. X-ray crystallographic analysis¹⁰⁾ revealed the configuration of **3a** which originates from the CC conformer of **2**. A computer generated perspective view of **3a** is shown in Fig 1. Transannular cyclization of the CC conformer was therefore selectively achieved. Selective hydrolysis of the 6-acetoxy group was easily performed by refluxing with Na_2CO_3 (1.1 eq) in MeOH-water to afford **3b**⁹⁾ (mp $126-7^\circ$) in 99% yield. Silylation ($t\text{BuMe}_2\text{SiCl}$, imidazole) and reduction (LiAlH_4 , THF, rt, 1 hr) of **3b** gave **3c**⁹⁾ (mp $85-6^\circ$, 96%). Bromide **3d**, obtained by treatment of **3c** with PBr_3 (in pyr, at $0^\circ \rightarrow \text{rt}$), was unstable and therefore immediately converted by reduction with LiEt_3BH (THF-ether, rt) into **3e**⁹⁾ (56% from **3c**). Demasking of **3e** ($n\text{Bu}_4\text{N}^+\text{F}^-$, THF, rt, 24 hr;

3f⁹): mp 94-5°, 91%) followed by Collins oxidation yielded (±) bicyclohumulene (4) (mp 73°, 88%, 30% from 2), whose nmr and ir spectra were superimposable with those of the natural product, isolated from *Plagiochila acanthophylla* subsp. *japonica*.^{5f)}

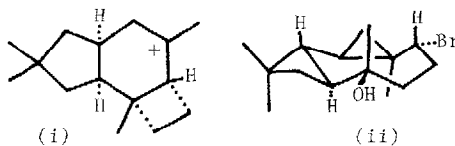
Another conformationally selective transannular cyclization of 2 was performed by treatment of 2 with trimethylsilyl trifluoromethanesulfonate¹¹⁾ (1.3 eq) in toluene at -20° → -10° for 6 hr and subsequent desilylation with K₂CO₃ in MeOH at rt for 20 hr. A mixture of alcohols 5a and 5b was obtained in 80% yield.¹²⁾ The mixture was separated by AgNO₃ impregnated silica gel chromatography (5a/5b = 1/2). A combination of shift reagent and decoupling experiments at 60 MHz and NMR spectrum at 270 MHz¹³⁾ established the configuration of olefins 5a¹⁴⁾ and 5b, both of which must be derived from the CT conformer of 2. The olefinic alcohol 5a was treated with PBr₃ (Et₂O, rt, 10 hr) to give a bromide 5c which was reduced without purification under Birch conditions (Na-*liq* NH₃, -78°, 30 min) to furnish an olefin 5d in 86% yield from 5a. The olefin 5d was quantitatively converted to a pair of stereoisomeric epoxides 5e and 5f (mCPBA, CH₂Cl₂, -15°, 10 min) from which epoxide 5e (moving faster on tlc) was separated in 55% yield. Treatment of 5e with LDA afforded an allylic alcohol 6 in 68% yield. Finally the allylic alcohol was hydrogenated (PtO₂, AcOEt, 0°) to an alcohol 7 (mp 47~8°, 92%, 8% from 2). Spectra of the synthetic alcohol 7 and the olefin 5d were superimposable respectively with those of natural products, africanol and africene, which were found in the marine animal, *Lemalia africana*.¹⁵⁾

In summary, two structurally different cyclohumulanoids have thus been synthesized from humulene epoxide 2 via conformationally selective transannular cyclization reactions. Conformational selectivity of the reaction could be rationalized by the different stabilities of transition states depending upon the presence (2 → 3a) or the absence (2 → 5a) of a nucleophile in the reaction medium because the epoxide 2 was shown to have same conformation in both the solvents namely Ac₂O and toluene as revealed by the ¹³C NMR spectral studies.

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References and Notes

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- 3) On treatment of humulene with NBS in aqueous acetone, there was obtained 3-hydroxy-6-bromo-tricyclohumulane (ii), whose conformation is known to be very similar to the original humulene CT conformer (X-ray analysis). (J.M. Greenwood, M.D. Solomon, J.K. Sutherland, A. Torre, *J. Chem. Soc., (C)*, 3004 (1968); F.H. Allen, D. Rogers, *Chem. Commun.*, 582 (1966); 588 (1967); *J. Chem. Soc., (B)*, 1047 (1968)). To date, no compound with the conformation resembling the humulene CC conformer has been recognized.



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- 5) Cyclohumulanoids such as illudoid^{a,2)}, hirsutanoid^{a,1)}, Pentalenanoids,^{b)} africanoid (africanol^{c)} and oxysenoxyrene^{d)}, and bicyclohumulene,^{f)} are all thought to originate from 9,10-dihydrohumulen-9-yl cation. (a) W.Parker, J.S.Roberts, R.Ramage, *Quart.Rev.* 21, 331 (1967). (b) D.E.Cane, T.Rossi, J.P.Pachlatko, *Tetrahedron Lett.*, 3639 (1979); H.Seto, H.Yonehara, *J.Antibiotics*, 33, 92 (1980). (c) B.Tursh, J.C.Braekman, D.Daloz, P.Pritz, A.Kelecom, R.Karlsson, D.Losman, *Tetrahedron Lett.*, 747 (1974). (d) Bohlmann^{e)} described oxysenoxyrene originated from humulene 2,3-epoxide, however, biosynthesis from 9,10-epoxide along a similar path to our synthesis seems to give a more reasonable explanation. (e) F.Bohlmann, C.Zdero, *Phytochem.* 17, 1669 (1978). (f) A.Matsuo, H.Nozaki, M.Nakayama, Y.Kushi, S.Hayashi, T.Komori, N.Kamijo, *J.Chem.Soc., Chem.Comm.* 174 (1979).
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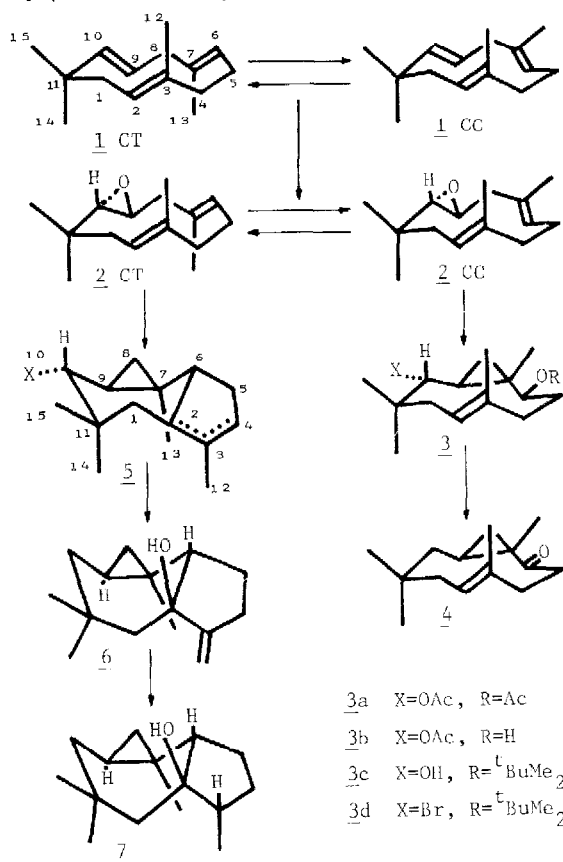
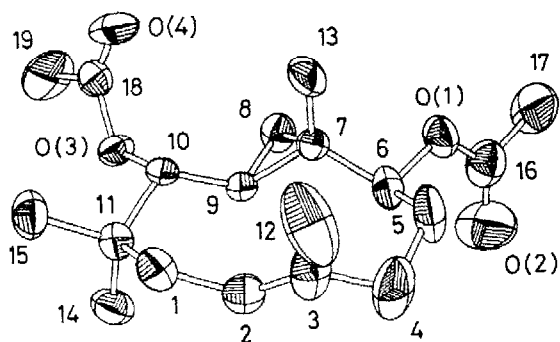


Fig 1 A perspective view of bicyclohumulenediol diacetate 3a (X-ray analysis)



- 3e X=H, R=^tBuMe₂Si
3f X=H, R=H
5c Δ^{2,3}, X=Br
5d Δ^{2,3}, X=H
5a Δ^{2,3}, X=OH
5e β-2,3-epoxy-, X=H
5b Δ^{3,4}, X=OH
5f α-2,3-epoxy-, X=H

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- 8) The conformation of humulene diepoxide is known to be very similar to that of humulene. M.E.Cradwick, P.D.Cradwick, G.A.Sim, J.Chem.Soc., Perkin II, 404 (1973).
- 9) Spectral data of all compounds are consistent with the structures depicted in the figure. NMR spectra were measured in CDCl_3 (3a, c, d, f) and in CCl_4 (5c, e, 6) and IR spectra were observed either in *nujol* or neat. NMR and IR spectral data are listed herein. 3a: IR 1722 cm^{-1} ; NMR 0.12 (1H, dd, $J=9.5, 4.5$, 8-H), 0.63 (1H, dd, $J=6, 4.5$, 8-H), 0.91, 1.08, 1.14 (each 3H, s, $\text{Me}\times 3$), 1.68 (3H, bs, 12-Me), 1.99 (6H, s, $\text{Ac}\times 2$), 4.10 (1H, dd, $J=9, 2.5$, 6-H), 4.52 (1H, d, $J=8, 10$ -H), 5.30 (1H, bd, $J=10.5, 2$ -H). 3c: IR 3450 cm^{-1} ; NMR 0.00 (6H, s, SiMe_2), 0.37 (1H, dd, $J=9.5, 4.5$, 8-H), 0.50 (1H, dd, $J=12, 4.5$, 8-H), 0.87 (9H, s, $^t\text{BuSi}$), 0.94 (3H, s, 13-Me), 1.04 (6H, s, $\text{Me}\times 2$), 1.63 (3H, bs, 12-Me), 2.71 (1H, dd, $J=9, 3, 6$ -H), 2.93 (1H, bd, $J=7.5, 10$ -H), 5.21 (1H, dd, $J=10.5, 3, 2$ -H). 3d: NMR 0.00 (6H, s, SiMe_2), 0.59 (1H, t, $J=5, 8$ -H), 0.86 (9H, s, $^t\text{BuSi}$), 1.00 (3H, s, 13-Me), 1.20 (3H, s, Me), 1.28 (3H, s, Me), 1.64 (3H, bs, 12-Me), 2.73 (1H, m, 6-H), 4.03 (1H, d, $J=8, 10$ -H), 5.2 (1H, m, 2-H). 3f: IR 3270 cm^{-1} ; NMR -0.17 (1H, dd, $J=4.5, 2.5, 8$ -H), 0.72 (1H, dd, $J=4.5, 3.5, 8$ -H), 0.91, 0.95, 1.11 (each 3H, s, $\text{Me}\times 3$), 1.66 (3H, bs, 12-Me), 2.74 (1H, dd, $J=8, 4, 6$ -H), 5.32 (1H, d, $J=10.5, 2$ -H). 5c: NMR 0.3~0.8 (3H, m, 8- $\text{H}_2, 9$ -H), 0.88 (3H, s, 13-Me), 1.04 (6H, s, $\text{Me}\times 2$), 1.63 (3H, bs, 12-Me), 3.74 (1H, d, $J=10, 10$ -H), 5e: NMR 0.05~0.7 (3H, m, 8- $\text{H}_2, 9$ -H), 0.87, 0.98, 1.02, 1.26 (each 3H, s, $\text{Me}\times 4$). 6: NMR 0.0~0.7 (3H, m, 8- $\text{H}_2, 9$ -H), 0.81, 0.98, 1.13, (each 3H, s, $\text{Me}\times 3$), 4.75 (1H, t, $J=2.5, 12$ -H), 4.95 (1H, t, $J=2.5, 12$ -H).
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- 12) Treatment of 2 with $\text{BF}_3\cdot\text{OEt}_2$ in ether leads to the formation of a similar mixture of the alcohols 5a and 5b(1:1) in 70% yield. J.A.Mlotkiewicz, J.Murray-Rust, P.Murray-Rust, W.Parker, F.G.Riddall, J.S.Roberts, A.Sattar, Tetrahedron Lett., 3887 (1979).
- 13) 5b: LIS ($\text{Eu}(\text{fod})_3$) S-values 18.78 (15-Me), 18.07 (9 αH), 10.96 (14-Me), 9.54 (8 αH), 6.29 (8 βH), 5.28 (13-Me), 2.95 (12-Me). NMR (270 MHz) 0.43 (1H, t, $J=4.5, 8\beta\text{H}$), 0.59 (1H, ddd, $J=8.5, 8, 4.5, 9\alpha\text{H}$), 0.74 (1H, dd, $J=8, 4.5, 8\alpha\text{H}$), 0.91 (1H, t, $J=13, 1\beta\text{H}$), 1.0 (3H, s, Me), 1.01 (6H, s, $\text{Me}\times 2$), 1.38 (1H, q, $J=9.3, 6\beta\text{H}$), 1.61 (3H, q, $J=1.3, 12$ -Me), 1.75 (1H, dd, $J=13.5, 3, 1\alpha\text{H}$), 2.11 (1H, m, 5 βH), 2.28 (1H, ddq, $J=15, 9.3, 2.5, 5\alpha\text{H}$), 2.56 (1H, m, 2 αH), 3.18 (1H, d, $J=8.5, 10\beta\text{H}$), 5.28 (1H, bs, 4-H).
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- 16) Address to which correspondence concerning the study should be sent.